



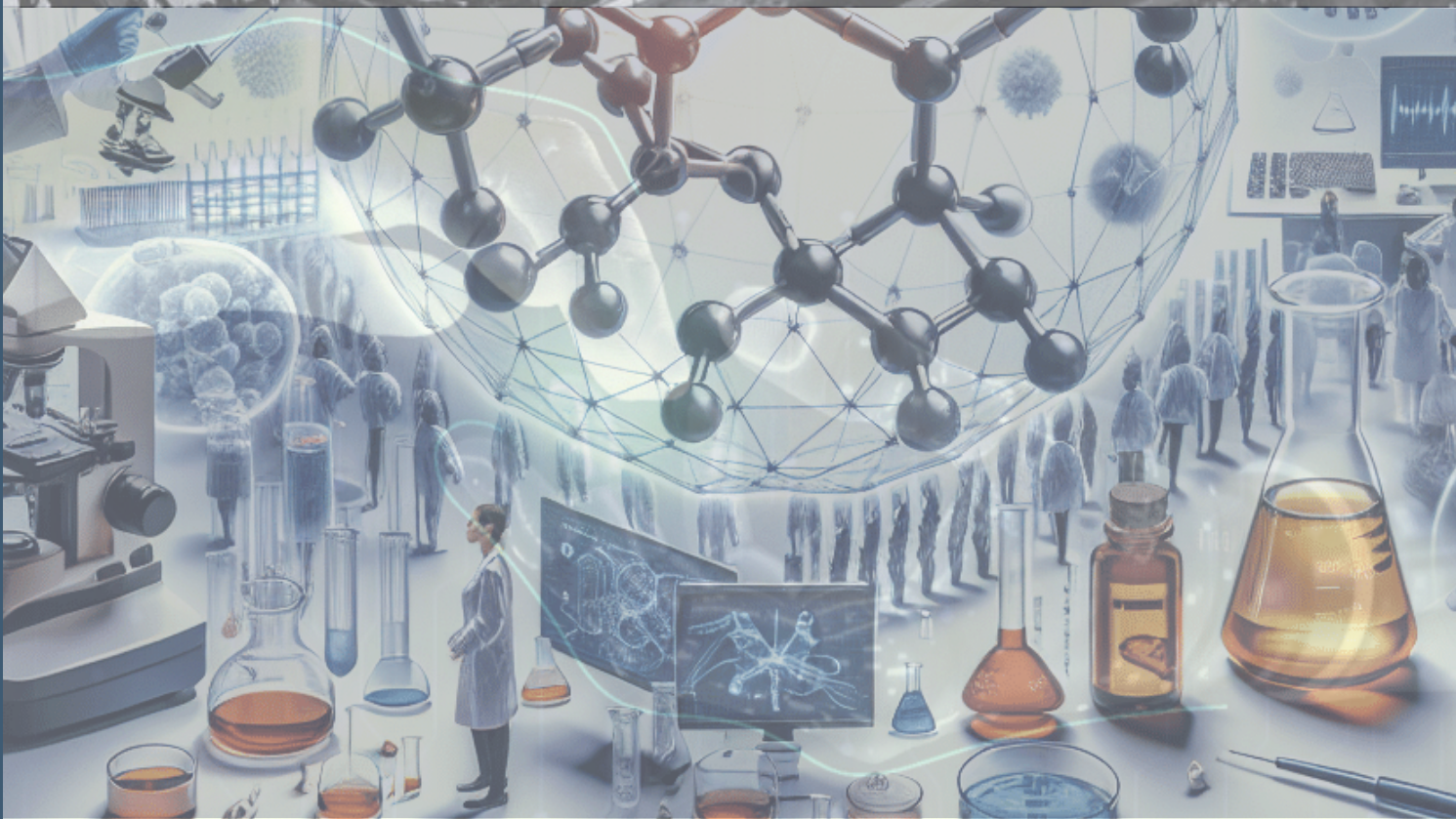
مؤتمر ريمار الدولي السابع للمعلوم الصرفة والتطبيقية

VII. International Rimar Congress of Pure, Applied Sciences

2025

FULL TEXT BOOK

RIMAR ACADEMY



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PREFACE

VII. International Rimar Congress of Pure, Applied Sciences was organized by Mardin University in collaboration with Rimar Academy. The primary objective of this event was to compile and disseminate valuable scientific knowledge and make a meaningful contribution to the future.

A substantial number of researchers from both local and international backgrounds demonstrated their interest in this conference. The scientific committee meticulously reviewed the submissions and ultimately accepted a select group of applicants—**15** in total—of whom **10** were approved by the scientific committee.

The core of this conference was the presentation of **9** full research papers, while the remaining articles and research findings are set to be featured in the coming issues of the MINAR Journal.

I would like to extend my sincere appreciation to all the contributors and scholars who played an essential role in making this conference a resounding success.

Your dedication and valuable contributions are deeply respected and acknowledged.

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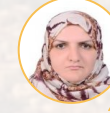
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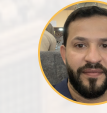
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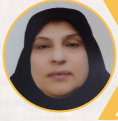


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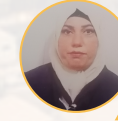
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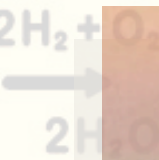
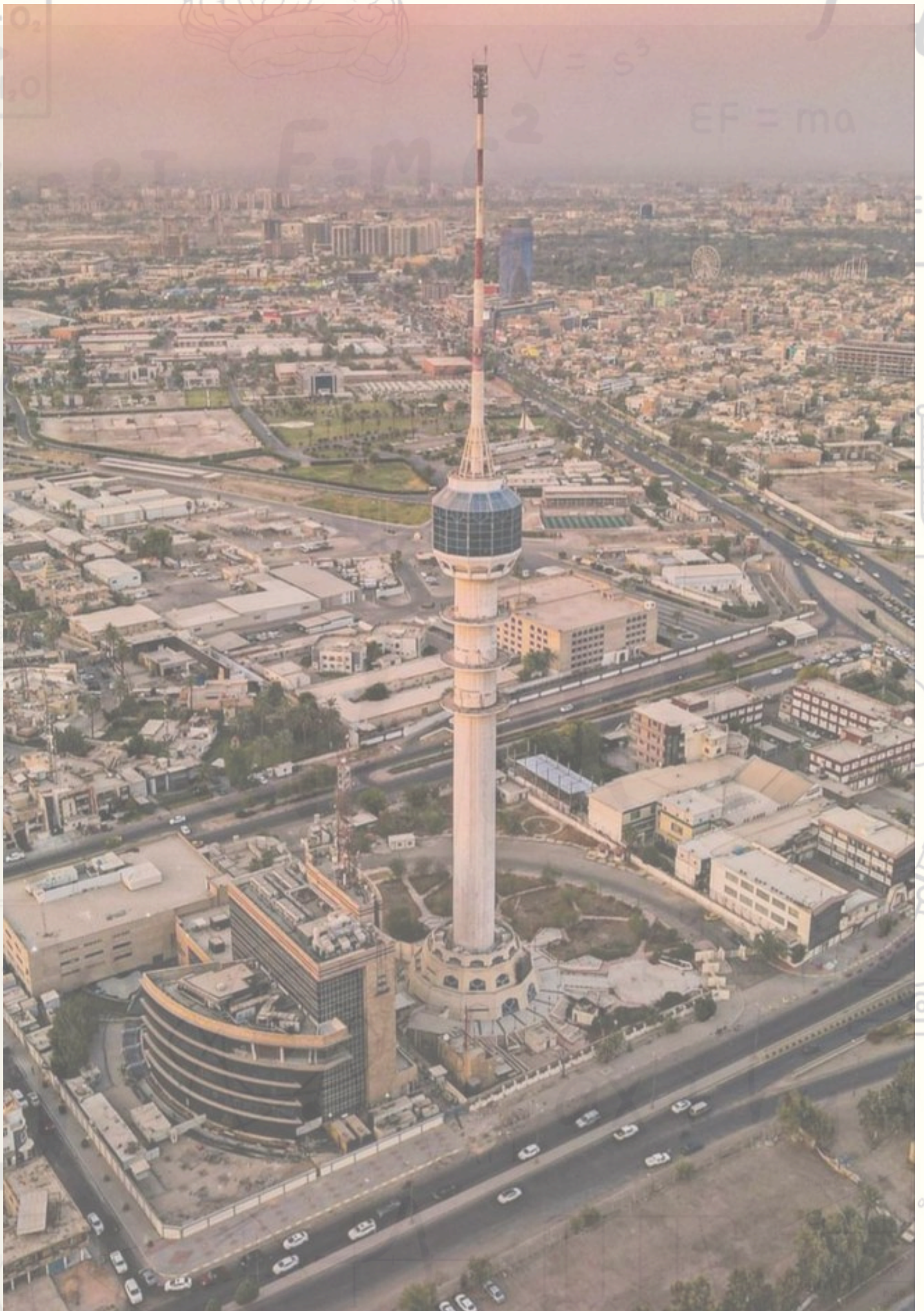
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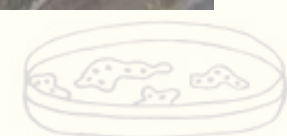
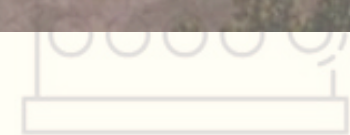


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Antileishmanial Effects of Multi-Walled Carbon Nanotubes and *Lactobacillus Acidophilus* Cell-Free Supernatants: A Comparative Study on *Leishmania Donovanii* and *Leishmania Tropica*

Mustafa Attiyah Hadid ¹

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Abstract:

MTT assay was used to evaluate antiparasitic effect of agents after preparing 4 concentrations (12.5, 25, 50, and 75 µg/ml) for each of multi-walled carbon nanotubes (MWCNTs), *Lactobacillus acidophilus* cell free supernatant (CFS), a combination of multi-walled carbon nanotubes and *Lactobacillus acidophilus* cell free supernatant (MWCNT+CFS), and pentostam as reference drug. Results showed that all treatments have antileishmanial impact when compared to untreated group. Most effective treatments were belonging to CFS against both parasites followed by their combinations with MWCNT. There was no synergistic impact of the combination compared to the individual treatments. In conclusion, CFS alone has the most antileishmanial impact and it is more potent than its combination with nanomaterials, while MWCNT showed less antileishmanial effect although still considerable compared to untreated group.

Keyword: *Antileishmanial, Cell viability, CFS, Lactobacillus, Leishmania donovani, Leishmania tropica, Multiwalled carbon nanotubes, Nanoparticles.*



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Introduction

Leishmaniasis is a neglected tropical illness that has been threatening the world for more than a century. Parasites belonging to the genus *Leishmania* are the cause of the disease. The parasite manifests in three ways: visceral leishmaniasis, cutaneous leishmaniasis, and mucocutaneous leishmaniasis (Sasidharan & Saudagar, 2023). Visceral leishmaniasis is the most fatal form among the three, caused by *Leishmania donovani* (Sasidharan et al., 2019). *Leishmania* has two life stages: the amastigote, which has round to oval bodies and is only found in the macrophages of infected vertebrate hosts, and the promastigote, which is motile and has a single anterior flagellum and is found in the gut of sand flies (Singh, 2006; Ul Bari & Rahman, 2008). The disease is transmitted among the host via the bite of a *Phlebotomus* sandfly. Between 70,000 and 1 million cases are reported globally, while 26,000 to 65,000 people die from the disease each year (Harrison & Atala, 2007; Rathinavel et al., 2021). Amphotericin B liposomal formulation is still the first line of treatment, despite the fact that several medications have been developed to date to prevent the condition. These medications have a number of disadvantages, including high toxicity, organ failures, non-specificity, and lack of economic viability (Sasidharan & Saudagar, 2023). Strong synthesis techniques are enabling the rapidly developing field of nanobiotechnology to make significant advances (Sasidharan & Saudagar, 2020). Researchers are interested in nanoscale particles because of their many biotechnological uses (Sasidharan & Saudagar, 2021, 2022). In the twenty-first century, carbon nanotubes (CNTs) have gained interest due to their potential uses in biosensors, electronics, tissue engineering, gene therapy, and neuro-engineering (Sasidharan & Saudagar, 2023). They fall into one of two categories: single-walled carbon nanotubes (CNTs) or multiple-walled CNTs (MWCNTs). CNTs composition, surface modification, target microbes, and reaction environment all affect how antimicrobial they are (Wang & Zhang, 2011). In addition to the physical process of biologically isolating microbial cells from their supportive environment, antimicrobial mechanisms of CNTs depend on the invasion of the microbial cell wall or membrane and the development of structural damages (Azizi-Lalabadi et al., 2020).

The advantages of CFS have been shown, in addition to the usage of probiotic microbes. For example, *in vitro*, the CFS of *Lactobacillus* species showed 100% antiparasitic activity against *Toxocara canis* (Walcher et al., 2023). The primary metabolites found in the CFS of bacteria belonging to the species *Lactobacillus* are organic acids, particularly lactic acid, acetic acid, proteins, and hydrogen peroxide (H₂O₂) (Abdul-Rahim et al., 2021; Belguesmia et al., 2020). The primary functions of lactic and acetic acids include lowering pH, regulating IL8, enhancing the host immune response, and inhibiting the development of pathogen resistance (Netto de Oliveira da Cunha et al., 2024).

This study was aimed to test the antiparasitic activity of MWCNTs, CFS of *L. acidophilus*, and their combinations against *L. donovani* and *L. tropica* as well as comparing the most antiparasitic effect on both parasites.

Materials and methods

Cell-free supernatant preparation

Lactobacillus acidophilus isolate was generously provided by Department of Biology/College of Science/University of Baghdad. The probiotic activated twice by subculturing the stock solution in De Man, Rogosa, and Sharpe (MRS) broth (Oxoid, UK) and incubated anaerobically at 37 °C for 24 hours.

To prepare cell-free supernatant (CFS), 120 ml of MRS broth with 1x10⁸ CFU/ml of 24 h grown *L. acidophilus* culture was incubated anaerobically at 37 °C, centrifuged at 4000 rpm for 15 m, and filtered through 0.22 µm pore-size filter papers (Microlab, UK) before lyophilising and storing at -20 °C (Hadid et al., 2025).

Functionalization of multi-walled carbon nanotubes (MWCNT)

A combination of 25% HNO₃ and 75% H₂SO₄ was used to treat 0.15 g of raw MWCNTs (SkySpring Nanomaterials, USA) in a flask. For 50 minutes at 37 °C, the raw material and acids were ultrasonically treated in the ultrasound bath. After that, the mixture was diluted with 40 ml of DI water using a 0.22 µm polypropylene membrane in a vacuum-filtered system. Next, 100% H₂O₂ was used in place of the solvent, and the same process was repeated by treating with H₂O₂ in a mild way. The particles were cleaned with deionised water until the filter's pH was about 7, and they were then dried in a furnace set between 170 and 200°C for 24 hours, producing functionalised MWCNT-COOH (MWCNTCOOH) (Ahmed, 2019)

Parasite Culture

The parasites *L. donovani* and *L. tropica* were graciously donated by the Biotechnology Research Centre at Al-Nahrain University. The promastigotes were cultured in 199 medium with 50 µg/ml penicillin and 10% HI-FBS at 26°C (Zghair et al., 2016).

MTT assay

Four concentrations (2.5, 25, 50 and 75 µg/ml) of each of MWCNT, CFS, MWCNT+CFS, and pentostam drug were prepared and dispersed in a flat-bottom 96-well microtiter plate containing *L. donovani* and *L. tropica* promastigotes (2 x 10⁵ parasites/well) and incubated at 25°C (Zghair et al., 2016). The combinations of MWCNT+CFS were made by adding MWCNT and CFS in 1:1 ratio for each concentration. The microtiter plate was incubated at 25°C for 24 hrs. After incubation, 10 µl of MTT solution was applied to each well. The microtiter plate was incubated for 4 hours at 25°C. To solubilise the formazan crystals, remove the medium and add 100µl of DMSO solution. After gently stirring the microtiter plate, it was left for fifteen minutes. The ELISA reader was used to measure the optical absorbance of these plates at 570 nm.

The cell viability was measured by the formula below:

$$\text{Viability (\%)} = (\text{Mean OD}_{\text{sample}} - \text{Mean OD}_{\text{blank}}) / (\text{Mean OD}_{\text{control}} - \text{Mean OD}_{\text{blank}}) \times 100$$

Statistical analysis

The data was analysed using ANOVA and Tukey's/Dunnett's multiple comparison tests, with a significance level of p-value < 0.05. These analyses were carried out with GraphPad Prism version 8.4.3.

Results

The IC₅₀ results revealed varied susceptibility of *L. donovani* and *L. tropica* to the investigated treatments (Figures 1 and 2). MWCNTs had modest activity against *L. donovani* (IC₅₀ = 24.14 µg/mL, R² = 0.84), while CFS had a significantly larger effect (IC₅₀ = 5.87 µg/mL, R² = 0.94). The combination of MWCNTs with CFS resulted in the same inhibitory concentration (IC₅₀ = 5.87 µg/mL, R² = 0.94), indicating no additional additive effect compared to CFS alone. Pentostam, the reference drug, showed the highest potency (IC₅₀ = 3.35 µg/mL, R² = 0.87).

MWCNTs showed slightly higher activity against *L. tropica* compared to *L. donovani* (IC₅₀ = 19.53 µg/mL, R² = 0.85). CFS showed substantial inhibition (IC₅₀ = 4.32 µg/mL, R² = 0.92). However, unlike *L. donovani*, the combination of MWCNTs + CFS demonstrated an intermediate impact (IC₅₀ = 12.43 µg/mL, R² = 0.87), which was weaker than CFS alone. Pentostam maintained its effectiveness (IC₅₀ = 3.97 µg/mL, R² = 0.91). Overall, these findings show that CFS outperforms MWCNTs in both *Leishmania* species, with Pentostam serving as the most effective control. Interestingly, while the combination did not outperform CFS alone in *L. donovani*, it had a lower efficacy in *L. tropica*, indicating species-specific interactions.

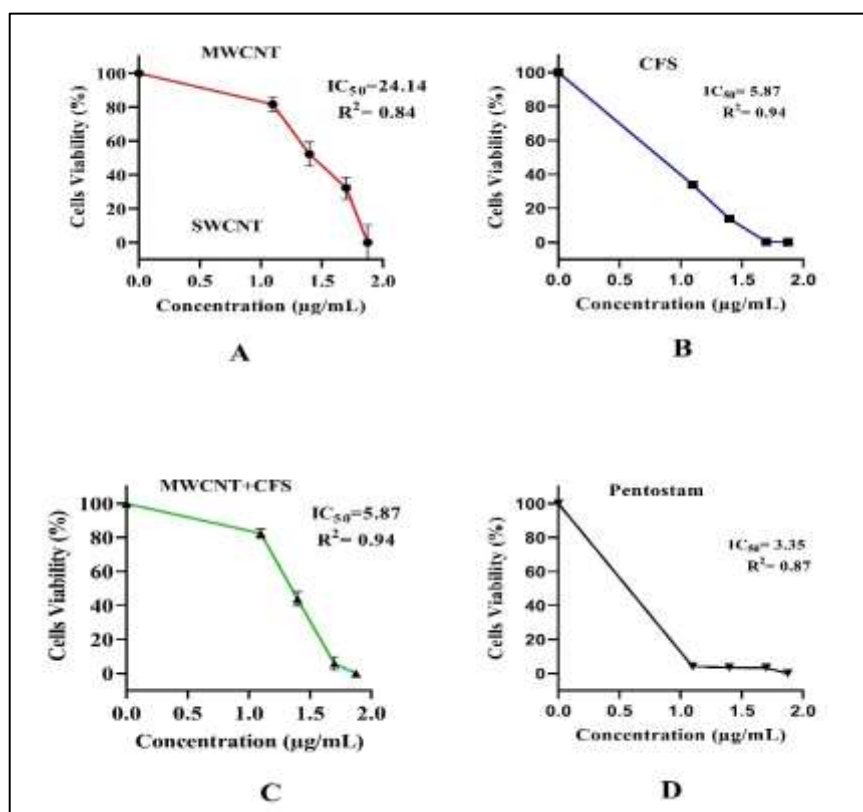


Figure 1 Antileishmanial impact of MWCNTs, CFS of *L. acidophilus* and their combinations against *Leishmania donovani*

Our results showed that all treatments were significant in reducing the parasite viability of both *L. donovani* and *L. tropica* when compared to untreated group (Figure 3). Also, the impact of CFS alone was higher against both parasites than nanomaterials alone nor their combination with CFS which is in line with IC₅₀ results. It is noteworthy that CFS treatment was more potent than pentostam against *L. donovani* and was as similar in potency as pentostam against *L. tropica*.

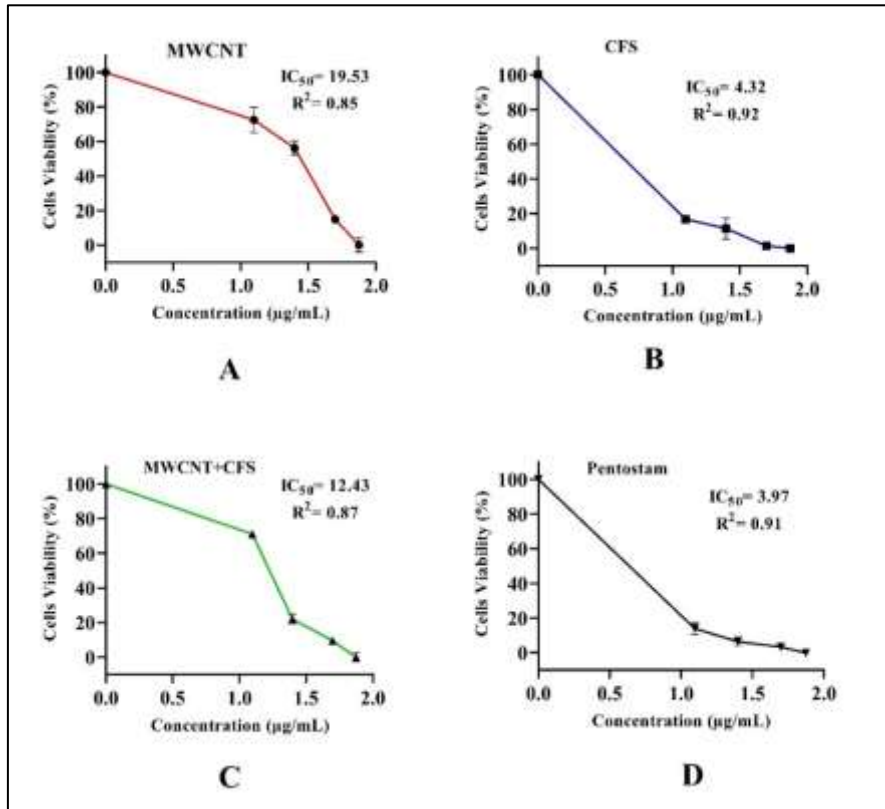


Figure 2 Antileishmanial impact of MWCNTs, CFS of *L. acidophilus* and their combinations against *Leishmania tropica*

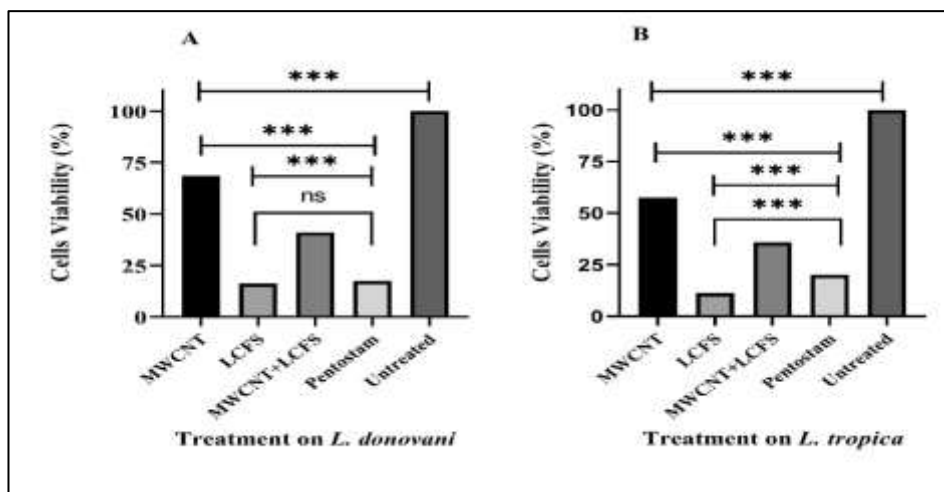


Figure 3 Antileishmanial effects of MWCNTs, *L. acidophilus* CFS, and their combination on (A) *Leishmania donovani* and (B) *Leishmania tropica*, with Pentostam included as a reference control

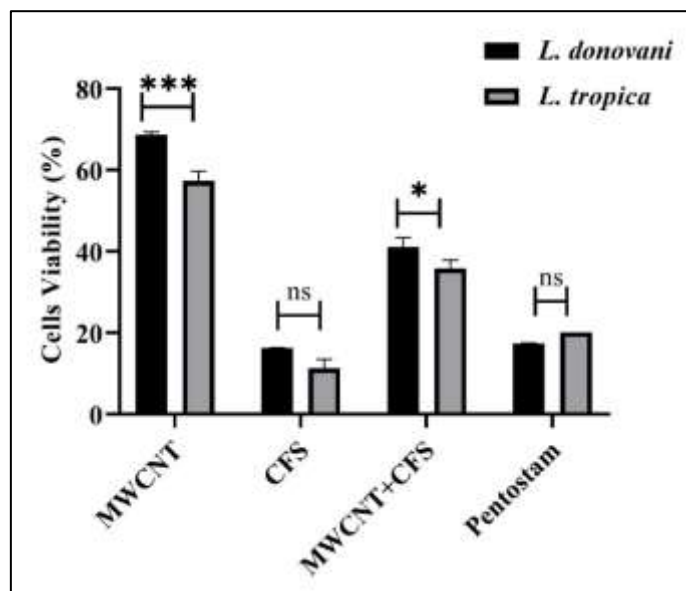


Figure 4 A comparison between the antileishmanial effect of MWCNTs, *L. acidophilus* CFS, and their combination toward *L. donovani* and *L. tropica*

When a comparison made between *L. donovani* and *L. tropica* results demonstrated that MWCNT was more effective against *L. tropica* than *L. donovani* similar findings obtained from MWCNT+CFS (Figure 5). On the other hand, there were no significant differences between *L. donovani* and *L. tropica* in the treatment of CFS alone and pentostam.

Discussions

This study examined the antileishmanial properties of *L. acidophilus* cell-free supernatant (CFS), multi-walled carbon nanotubes (MWCNTs), and their combination against *L. tropica* and *L. donovani*. The findings showed clear species-specific reactions, with pentostam acting as the most successful reference control and CFS continuously exhibiting higher activity than MWCNTs.

With IC_{50} values that were comparable to pentostam, CFS demonstrated impressive inhibitory effects on both *L. donovani* and *L. tropica*. Probiotic-derived metabolites have been shown in recent research to have antiparasitic potential, producing bacteriocins, organic acids, and biosurfactants with direct antiparasitic activity. The way probiotics work is by secreting organic acids like lactic, butyric, and acetic acid as well as antibacterial substances like bacteriocins that may have antiparasitic properties (Chiodo et al., 2010; Hernández-González et al., 2021; LeBlanc et al., 2017).

Our results further highlight the potential of probiotics and their metabolites as supplemental or alternative approaches to the treatment of leishmaniasis. It's interesting to note that CFS activity was much stronger in *L. donovani* than in pentostam, suggesting that it might be better than traditional medications in some circumstances.

On the other hand, the antiparasitic effects of MWCNTs were mild but noteworthy. Despite generally being less effective than CFS, MWCNTs demonstrated their potential as antiparasitic drugs by significantly reducing parasite viability as compared to untreated groups. In *L. donovani*, for example, parasite inhibition was close to 40%, but in *L. tropica*, it

was over 50%. According to this, carbon nanotubes may have significant antiparasitic effects. They may work by disrupting membranes, inducing oxidative stress, and interfering with parasite metabolic processes (Demir, 2022). A study showed that amphotericin B functionalised carbon nanotubes had no discernible deleterious effects and were far more effective against leishmaniasis than amphotericin B (Foldvari & Bagonluri, 2008). These findings matched with earlier research that shown the effectiveness of CNTs against *Leishmania* species *in vitro* (Mohebbali et al., 2009).

MWCNTs have significant potential as an antiparasitic agent, even if CFS seems more effective and secure. This is especially true if future studies concentrate on functionalisation or tailored delivery methods to increase selectivity.

The effects of combining the two agents varied according on the type of parasite. The combination of MWCNT and CFS in *L. donovani* had an effect that was comparable to that of CFS alone, indicating no additive or synergistic advantage. The combination, however, was less efficient in *L. tropica* than CFS alone, possibly as a result of surface interference between the nanomaterials and the active metabolites in the supernatant.

These results demonstrate the complex nature of interactions between nanomaterials and biomolecules and the necessity of carefully assessing combinations rather than relying on the assumption that they will increase efficacy. In this regard (Auría-Soro et al., 2019) mentioned that nanoparticles can behave differently depending on the surrounding medium conditions (pH, ionic strength, organic matter, etc.), which may lead to aggregation or interaction with other biomolecules. Such changes can create heterogeneous morphologies and reduce stability or biocompatibility.

MWCNTs and the MWCNT + CFS combination were dramatically more effective against *L. tropica* than *L. donovani*; however, there was no difference in CFS or pentostam activity between the two species. This implies that the susceptibility of *Leishmania* spp. to nanomaterial-based therapies may be strain- or species-dependent.

Conclusions

L. acidophilus cell-free supernatants (CFS) exhibited strong antileishmanial activity against *L. donovani* and *L. tropica*, in some cases comparable to the reference drug pentostam. MWCNTs established considerable inhibitory effects as compared to untreated controls, showing their antiparasitic potential even though they were not as effective as CFS. MWCNTs and CFS produced species-specific results, with *L. donovani* showing no extra antiparasitic impact and *L. tropica* showing decreased effect. These results show that CFS is a promising natural treatment option, but they also imply that more optimisation of nanomaterials is necessary to maximise their antileishmanial potential.

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Antimicrobial Evaluation of Chemically Synthesized ZnO Nanoparticles Against Escherichia Coli

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Abstract:

MTT assay was used to evaluate antiparasitic effect of agents after preparing 4 concentrations (12.5, 25, 50, 100). This study investigates the antimicrobial activity of zinc oxide (ZnO) nanoparticles synthesized by chemical methods at concentration of 11.32×10^{-1} M against Escherichia coli (E. coli). The ZnO nanoparticles characterized by X-ray diffraction (XRD), Field Emission Scanning Electron Microscopy (FE-SEM), and UV-visible spectroscopy to confirm their nanoscale size, morphology, and optical properties.

The results revealed a clear inhibition zone around ZnO nanoparticles, demonstrating significant antibacterial activity. Chemically synthesized ZnO nanoparticles show promising potential as effective antimicrobial agents against E. coli, suggesting their possible application in biomedical, pharmaceutical, and environmental fields.

Keywords: Zinc Oxide Nanoparticles, Antibacterial Activity, Escherichia Coli, Biomedical Applications, Nanotechnology.



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Introduction

Nanotechnology's wide range of inventive uses in advanced textile manufacturing, food processing, medical diagnostics and therapeutics, and agricultural production have made it one of the most forward-thinking and quickly growing areas of contemporary scientific and technological research [1,2]. In order to define, optimize, and improve the structural and functional properties of both natural and artificial systems, this field uses extremely sophisticated approaches [3].

Materials that have innovative and significantly improved physicochemical and biological properties, as well as special phenomena and functionalities that are not present in their bulk equivalents, are the subject of the current investigation because of their nanoscale dimensions [4]. The surface-to-volume ratios of nanoparticles (NPs), which are generally between 1 and 100 nm in size, are significantly larger than those of macroscale materials [5–7]. These unique structural characteristics give nanoparticles remarkable physicochemical characteristics, which make them extremely important in a variety of applications, including the development of sophisticated biosensors and nanomedicine [8].

When metallic nanoparticles, such as silver (Ag), zinc oxide (ZnO), and titanium dioxide (TiO₂), are reduced to the nanoscale level, they can be engineered to exhibit tunable characteristics, including size, morphology, crystallinity, composition, and various physicochemical properties (chemical, mechanical, electrical, structural, morphological, and optical) [9]. For biomedical applications, this nanoscale alteration is essential because it permits special interactions with biological molecules and makes it easier for nanoparticles to enter cellular structures [8]. Nanostructured materials' higher percentage of surface atoms improves their reactivity and interaction potential, which helps explain why bio nanotechnology and the basic and applied sciences are becoming more and more interested in them. Zinc oxide (ZnO), one of the metal oxide nanoparticles, has garnered a lot of scientific attention because of its remarkable antibacterial activity against a variety of bacterial species [10–15]. Comparative studies have demonstrated that when ZnO is produced at the nanoscale, it has strong antibacterial properties. ZnO nanoparticles start a variety of bactericidal processes when they come into contact with bacterial membranes or intracellular components [14,16].

ZnO nanoparticles' high toxicity to bacteria and compatibility with human cells have led to their growing application as antimicrobial agents in the food sector. Numerous studies have verified ZnO-NPs' biocompatibility and biosafety, showing that they are strong antibacterial agents without harming human cells [17–19]. The effectiveness of nanoparticles against bacteria is largely dependent on their high surface area-to-volume ratio [20]. More investigation into ZnO nanoparticles' antibacterial potential is crucial for developing global health solutions and battling microbial resistance because of their unique physicochemical properties [21].

Experimental

Escherichia coli were selected and ready for testing in the current bioactivity application. *Escherichia coli* were cultivated for a whole day in nutrient broth. After being incubated for the entire night, the isolates were diluted to the McFarland standard and 0.1 ml of a diluted bacterial culture was uniformly spread on Muller-Hinton agar plates.

Powdered ZnO nanoparticles were dissolved in deionized water to create solutions with concentrations of ZnO nanoparticles of 11.32×10^{-1} M, homogenized by a vortex mixer, and added to the plates. Growth inhibitory zone formation was examined following a final 24-hour incubation period at 37 °C.

Results and discussion

The XRD patterns of ZnO NPs are given as indicated in figure 1 in order to investigate the phase structure for the ZnO nanoparticles that were synthesized as powder. The ZnO nanoparticles' polycrystalline structure is revealed by the X-ray pattern, which shows 11 peaks at the following diffraction angles: 31.8° (100), 34.4° (002), 36.2° (101), 47.6° (102), 56.6° (110), 63.3° (103), 66.6° (200), 68.3° (112), 69.2° (004), 72.8° (201), and 77.2° (202). A hexagonal structure is mentioned in [22–23].

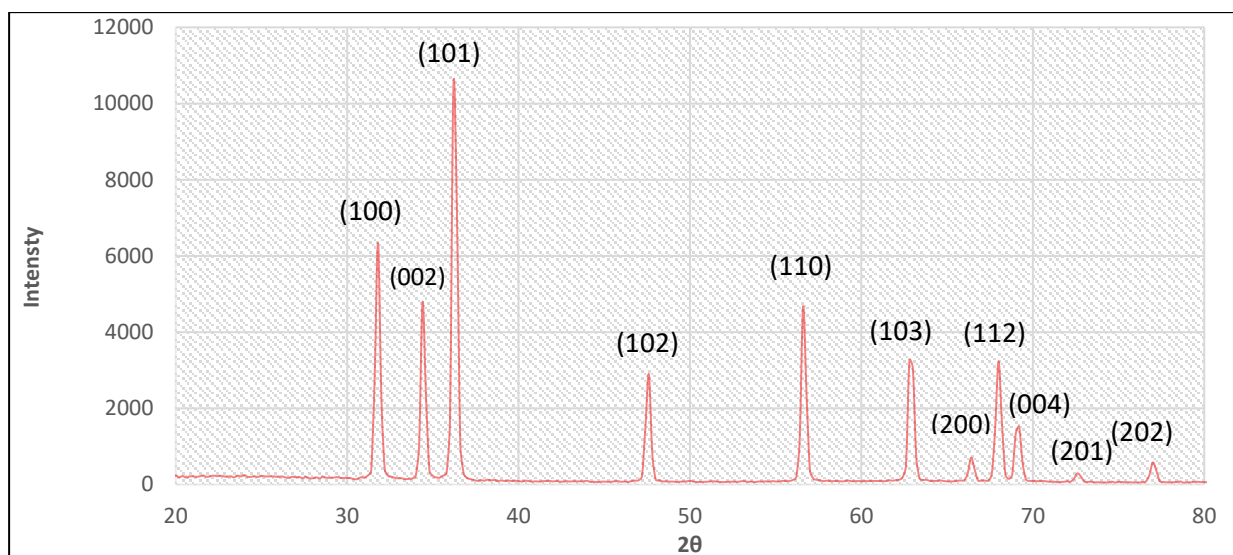


Figure 1: X-ray Diffraction of ZnO Nanoparticles

Field Emission- Scanning Electron Microscopy plays a crucial role in numerous applications and is a practical way to examine the structure of sample. Figure 2 shows FE-SEM pictures of ZnO nanoparticle, with the surface coated with equally distributed particles of different types (nanosheets, star shapes, and rods).

A uniform distribution of nanoparticles could be observed by analyzing the image in Figure 2. The distribution of particle sizes progressively shifts from smaller to larger aggregates. According to the particle size value determined by FE-SEM measurements, the size ranges from 26.23 to 379.3 nm at a concentration of 11.32×10^{-1} M. These results

Element	Atomic %	Atomic % Error	Weight %	Weight % Error
C	14.9	0.7	4.6	0.2
O	36.9	0.7	15.1	0.3
Zn	48.2	0.5	80.4	0.8

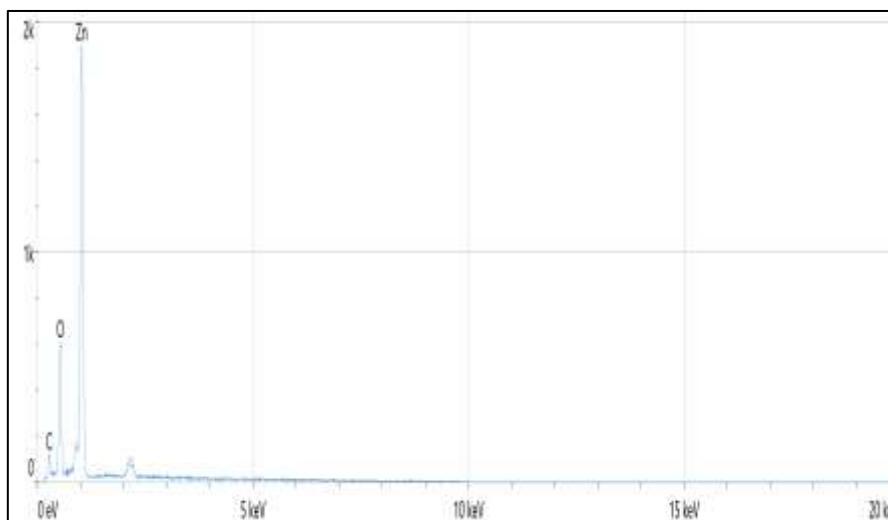


Figure 5: The EDX spectrum of ZnO Nanoparticles.

The absorbance of ZnO colloidal particles was measured, as shown in Figure (5). The peak was found to be in the 378–386 nm range, and the absorption in this area is caused by the development of zinc oxide nanoparticles [26].

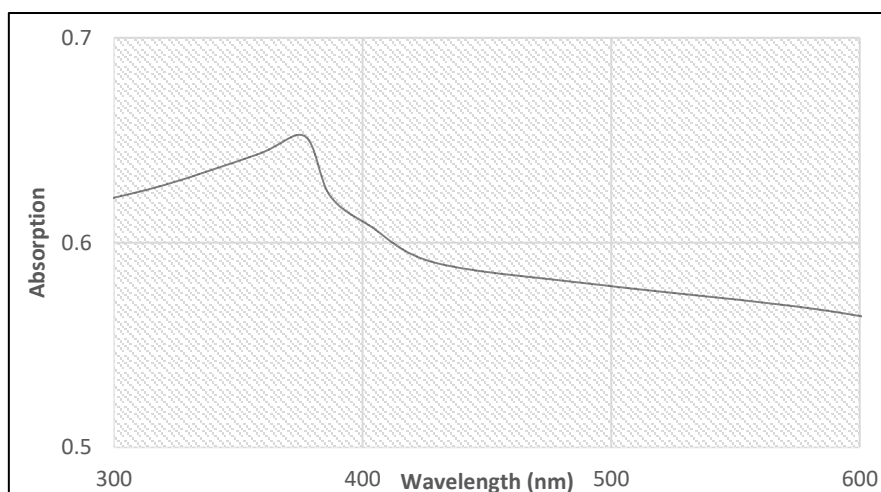


Figure 5: The absorption spectra of ZnO Nanoparticles .

Figure 6 shows *Escherichia coli* bacteria isolates before treatment and the antibacterial activity of ZnO nanoparticles against *Escherichia coli*. Heat-induced free radical production is linked to ZnO nanoparticle activity; when these radicals contact with the bacterial cell membrane, peroxidation results. Bacteria can be broken down and eliminated as a result of the oxidation of organic substances by the ensuing highly active oxygen molecules. The

findings demonstrated that the antibacterial properties of ZnO nanoparticles at 11.32×10^{-1} M were superior to those.

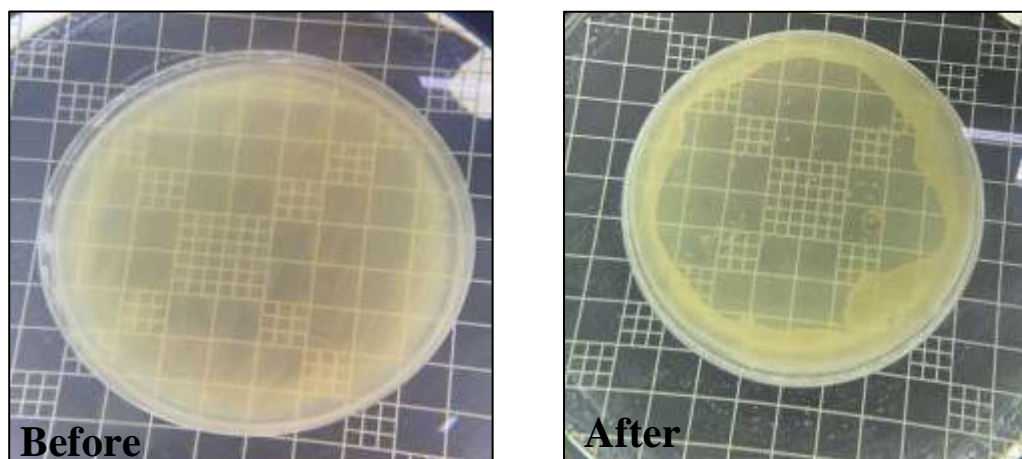


Figure 6: Before and after antibacterial activity of ZnO nanoparticles against Escherichia coli.

Conclusions

The results demonstrate that chemically produced ZnO NPs have a great deal of promise as strong antibacterial agents, especially when it comes to Gram-negative bacteria like *E. coli*. Therefore, in applications related to biomedicine, the environment, and food preservation, ZnO nanoparticles can be regarded as a promising substitute for traditional antibiotics. It is advised that more research be done to examine their cytotoxicity, biocompatibility, and functionality in intricate biological systems.

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Theoretical Investigation of Shape Coexistence and Ground-State Deformation in Neutron-Deficient Mercury Isotopes

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Abstract:

using Skyrme-Hartree-Fock-Bogoliubov (SHFB) calculations implemented in v3.00 of the HFBTHO code. The approach for this study used the full SLy4 interaction with a quasiparticle energy cutoff of 60 MeV for both neutrons and protons. The basis space was increased to 20 harmonic oscillator shells to achieve maximum computational accuracy, and the pairing strength for neutrons was set to 300 MeV. For odd-mass isotopes (e.g. N=103), the quasiparticle blocking method was used to accurately calculate the potential energy surface. The quasiparticle blocking and constrained HFB calculations used the quadrupole moment (Q_{20}) as a collective coordinate for potential-energy-surface mapping, clearly demonstrating Shape Coexistence between oblate and prolate shapes in isotopes near the mid-neutron shell ($N \approx 104$). The results from isotopes with neutron number (N=106) or lower show a high degree of agreement between experimental measurements of binding energy, separation energy and deformation parameters and calculated ground state configuration, providing strong evidence that the SHFB model has successfully predicted the complex dynamical interactions between shell structure and geometrical deformations in the chain of Mercury isotopes.

Keywords: *Shape Coexistence - Mercury Isotopes - Nuclear Deformation - Potential Energy Surface.*



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1. Introduction

A fundamental challenge is that the internal structure of the atomic nucleus remains a complex, multi-body quantum system, representing a major challenge in nuclear physics where the properties of the nucleus arise from complex interactions between individual particle degrees of freedom. In which each neutron or proton moves in a specific orbit, there are also collective degrees of freedom, which describe the interconnected collective motion of nucleons, also causing phenomena such as nuclear vibration and spin [1]. One of the most prominent manifestations of this interaction is the phenomenon of shape coexistence, in which different nuclear states with highly contrasting geometries (such as spherical, elongated "cigar-shaped," or flattened "disc-shaped") exist at similar energies, these shapes may even compete to become the fundamental state of the nucleus, challenging simple single-shape models [2,3]. The origin of this phenomenon lies in the structure of the nuclear shell, closing the shells at specific "magic numbers" provides additional stability to the nucleus [4]. Near these closed shells, the excitation of a few nucleons ("valence particles") across the energy gaps (forming "extraneous states") can push the nucleus toward a highly distorted shape that coexists with its original near-spherical form [5], the coupling of nucleons in pairs plays a crucial role, as coupling tends to stabilize spherical or weakly distorted shapes and thus directly competes with the distortion effects resulting from single-particle orbitals, the balance between these two forces determines the final nuclear shape [6]. The classic study of mercury isotopes is a prime example of neutron-deficient coexistence (below the closed shell of the proton $Z=82$), in these isotopes, a weakly distorted near-spherical ground state competes with highly oblate open states resulting from the excitation of protons across the closed shell $Z=82$ [7]. The theoretical tool used to describe these phenomena is the consistent mean field. Self-consistent mean field theories, such as the Hartree-Fock-Bogoliubov (HFB) method, employ effective Skyrme interactions, which allow for the simultaneous calculation of nucleus deformation and coupling properties. By calculating potential energy surfaces, the ground state, co-states, and structural evolution across different isotope series can be determined [8]. Nucleon-Nucleon interactions and the principles of structural evolution in nuclei far from stability are crucially revealed by changes in nuclear parameters including radii, charge distributions, and density profiles as well as the appearance of nuclear deformation phenomena [9-10]. The aim of this work is to conduct a systematic theoretical study using Skyrme-Hartree-Fock-Bogoliubov (SHFB) calculations of the deformation and co-state evolution in the low-neutron mercury isotope series ($A = 172-190$) by analyzing potential energy surfaces, deformations, and binding energies to understand the interaction between single-particle effects, coupling, and collective motion.

2. Methodology

Self-consistent nuclear Density Functional Theory (DFT) has been used as a theoretical tool to explore the nuclear structure of neutron-deficient mercury isotopes. The Skyrme Hartree-Fock-Bogoliubov (HFB), as the main theoretical framework for this study, provides a microscopic and comprehensive model that describes the properties of open shell nuclei (with particle-holes) and superfluidity (particle-particle) in a similar manner. An accurate

description of the ground-state properties of nuclei located far away from the line of stability, where deformation and pairing correlations have an important influence, is required for a satisfactory description of the nuclear structure of these territories.

2.1 Governing Theoretical Model:

The HFB theory extends the Hartree-Fock limit from the HFB model, including the pairing correlations in a self-consistent way with the help of the Bogoliubov-Valatin transformation, and defines $|\Phi\rangle$ as the state of the nucleus with respect to independent quasiparticles. The quasiparticles consist of superpositions of the creation and annihilation operators, which we denote as c^\dagger and c , respectively, defined through the following linear transformation:

$$\beta_k^\dagger = \sum_l (U_{lk} c_l^\dagger + V_{lk} c_l) \quad (1)$$

$$\beta_k = \sum_l (U_{lk}^* c_l + V_{lk}^* c_l^\dagger) \quad (2)$$

In this case, the indices l , and k indicate the single-particle basis states. The coefficients U_{lk} , and V_{lk} make up the Hartree-Fock and Bogoliubov Wavefunction, and fulfill the unitary condition such that the quasiparticle operators operate under the fermion commutation relations [11].

The two main quantities used in the HFB formalism are the normal density matrix ρ and the pairing tensor κ , which represent the particle-hole and particle-particle correlation, respectively, and are expressed in terms of the Bogoliubov coefficients:

$$\rho_{lm} = \langle \Phi | c_m^\dagger c_l | \Phi \rangle = (V * V^T)_{lm} \quad (3)$$

$$\kappa_{lm} = \langle \Phi | c_m c_l | \Phi \rangle = (V * U^T)_{lm} \quad (4)$$

The total binding energy of a nucleus can therefore be represented as a functional of the various densities. In the context of Skyrme Density Functional Theory, the binding energy functional will include contributions from the kinetic energy, the Skyrme interaction (coming from the effective nucleon-nucleon force), the Coulomb interaction and the pairing energy according to the equations:

$$E[\rho, \kappa] = E_{kin[\rho]} + E_{Skyrme[\rho]} + E_{Coulomb[\rho]} + E_{Pairing[\kappa]} \quad (5)$$

To arrive at the ground state, we can minimize the associated energy functional with respect to densities ρ and κ while preserving the constraint of constant total number of particles in the system. Thus, in regard to the variational principle, we can implement the method of Lagrange multipliers and apply the associated multiplier chemical potential λ [12]. The variational formulation can then be stated as follows:

$$\delta \langle \Phi | \hat{H} - \lambda \hat{N} | \Phi \rangle = 0 \quad (6)$$

When applying the method of the HFB equations to determine both the quasiparticle energies (E_k) and the wave functions (U , V), the resulting equations in coordinate space will be analogous to the coupled differential equations that define the system's quasiparticle states.

$$\Sigma_l(h_{il} - \lambda \delta_{il})U_{lk} + \Sigma_l \Delta_{il} V_{lk} = E_k U_{lk} \quad (7)$$

$$\Sigma_l \Delta_{il}^* U_{lk} - \Sigma_l (h_{il}^* - \lambda \delta_{il}) V_{lk} = E_k V_{lk} \quad (8)$$

In the equations below, h_{il} is the mean-field Hamiltonian (the Hartree-Fock field). This mean-field Hamiltonian includes both the kinetic energy of the i -th nucleon and the potential Γ , which arises from the interaction of this nucleon with the other nucleons in the system.

$$h_{il} = t_{il} + \Gamma_{il[\rho]} \quad (9)$$

The term Δ_{il} is the pairing potential (or pairing field), which characterizes the superfluid correlations and is dependent on the anomalous density κ :

$$\Delta_{il} = \left(\frac{1}{2}\right) \Sigma_{pq} \tilde{v}_{ilpq} \kappa_{pq} \quad (10)$$

From these equations, we can see that effective interaction v couples the mean field Γ with the pairing field Δ [13]. The size of the pairing gap derived from Δ dictates the nuclear response to single particle excitations, so it is also critical when determining the energy associated with shape coexistence as well as the binding energies of all isotopes in the mercury chain.

The solution to the HFB equations comes about by way of a self-consistent iterative approach; this begins with an approximate (educated) assumption for the densities ρ and κ , which in turn create h and Δ via the potentials. Next, we solve the eigenvalue equations in order to generate the new quasi-particle wave functions (U, V) and wave function energies (E_k). We can then generate new densities [14]. This cycle continues until the total energy and densities converge within a predetermined allowable numerical error, allowing the mean fields and wavefunctions producing those mean fields to be self-consistent. Additionally, we will need to adjust the chemical potentials λ_n and λ_p at each iteration to assure that the expectation value of the density operators matches the actual neutron (N) and proton (Z) populations present in the nucleus:

$$\langle \bar{N} \rangle = Tr(\rho_n) = N; \langle \bar{Z} \rangle = Tr(\rho_p) = Z \quad (11)$$

This rigorous mathematical framework allows for the simultaneous determination of nuclear deformation, radii, and separation energies, providing a robust theoretical basis for interpreting experimental data in the neutron-deficient region [15].

2.2 Energy Density Functional and Pairing Interaction:

The choice of the energy density functional is critical as it effectively parameterizes the underlying nuclear forces, in this work, the SLY4 (Skyrme-Lyon) parameterization was selected for the particle-hole (mean-field) part of the functional [16]. SLY4 is a well-calibrated functional known for providing a reliable description of nuclear matter properties, binding energies, charge radii across the nuclear chart, and particularly for its performance in describing neutron-rich systems and fission barriers.

$$V_{pair} = V_0 \left[1 - \eta \left(\frac{\rho(r)}{\rho_0} \right)^\gamma \right] \delta(r_1 - r_2) \quad (12)$$

For the particle-particle (pairing) channel, a separate effective interaction is required. We employed the Density-Dependent Delta Interaction (DDDI) [17], this is a zero-range force whose strength is modulated by the local nucleon density $\rho(r)$, this density dependence effectively confines the pairing interaction to the nuclear surface, suppressing it in the high-density interior, which is a more physically realistic scenario, the functional form and parameters used are summarized in Table 1.

Table 1: Parameters of the SLY4 Skyrme Functional and the DDDI Pairing Interaction.

Parameter	Value	Unit	Description
SLY4 Functional			
t_0	-2488.91	MeV fm ³	Zero-range term
t_1	486.82	MeV fm ⁵	Momentum-dependent term
t_2	-546.39	MeV fm ⁵	Momentum-dependent term
t_3	13777.0	MeV fm ^{3+3α}	Density-dependent term
x_0	0.834	dimensionless	Spin-exchange parameter
x_1	-0.344	dimensionless	Spin-exchange parameter
x_2	-1.000	dimensionless	Spin-exchange parameter
x_3	1.354	dimensionless	Spin-exchange parameter
α	1/6	dimensionless	Density-dependence exponent
W_0	123.0	MeV fm ⁵	Spin-orbit strength
DDDI Pairing			
V_0	300.0	MeV fm ³	Bare pairing strength
η	1.0	dimensionless	Density dependence type (surface)
γ	1.0	dimensionless	Density dependence exponent
ρ_0	0.16	fm ⁻³	Saturation density of nuclear matter

The pairing strength V_0 was fine-tuned to reproduce the general trend of the odd-even mass staggering in this specific mass region [18]. The pairing calculations are performed within a defined energy window, where all quasiparticle states up to an energy cutoff are included in the pairing active space [19].

2.3 Computational Procedure and Numerical Details:

All calculations were executed using the solver code HFBTHO (version 3.00) [20], this code solves the axial HFB equations by expanding the quasiparticle wave duty on a transformed harmonic oscillator (HO) basis, to investigate shape coexistence, constrained calculations were performed to generate the Potential Energy Surface (PES) for each nucleus, this was achieved by adding a quadratic penalty constraint to the energy functional, forcing the expectation value of the axial quadrupole moment operator, $\langle Q_{20} \rangle$, to a desired value [21], by systematically scanning a range of Q_{20} values, the total energy as a function of deformation was mapped.

For odd-A isotopes, the unpaired nucleon was treated using the blocking approximation[22], this involves performing a separate HFB calculation for each single-quasiparticle configuration near the Fermi surface, the state of the odd nucleon is "blocked," meaning it is excluded from the scattering processes that contribute to pairing, the ground state of the odd-A nucleus is then identified as the blocked configuration with the lowest total energy [23-24]. Table 2, shows the numerical parameter values and the arithmetic setup.

Table 2: Numerical Parameters and Computational Setup.

Parameter	Value / Setting	Description
Solver Code	HFBTHO (v3.00)	Solves axial HFB equations in an HO basis.
Basis Type	Deformed Harmonic Oscillator	Basis adapted to nuclear deformation.
Basis Size (N_max)	20	Number of major HO shells included.
Oscillator Length (b)	2.2 fm	Length parameter of the HO basis.
Quasiparticle Energy Cutoff	60 MeV	Energy window for pairing active space.
Constraint Operator	Axial Quadrupole Moment (Q_{20})	Used to map the Potential Energy Surface.
Odd-A Nuclei Treatment	Blocking Approximation	Accounts for the unpaired nucleon.
Corrections Applied	Center-of-Mass, Rotational Energy	Approximate restoration of broken symmetries.
Convergence Criterion	1×10^{-6} MeV	Required precision on the total energy for self-consistency.

3. Results and Discussion

3.1 Binding Energy Systematics:

A binding energy for a nucleus is typically calculated by dividing the binding energy (B) by the total number of nucleons (A) i.e. (B/A). A long-standing concept of how a nucleus is stable is by its binding energy (B/A) per nucleon or an average binding energy per nucleon. The binding energy (B/A), will measure how stable a given nucleus is relative to other nuclei across the spectrum of isotopes through the arrangement of nucleons, the pairing of nucleons, and a nuclear deformation.

Using the HFB formalism, all isotopes that constitute the isotopic chain that is being considered, exhibit parabolic behaviour. At the neutron mid-shell (where $N=105$ and $A \approx 184-186$), there is a local maximum for the (B/A) values (binding energy per nucleon) resulting from the saturation effect in deformed nuclei, which allows each nucleon to collectively interact and behave in a band-like manner. The results from the HFB calculations align with experimental results, which provide further evidence for the physical validity of the HFB

formalism and the energy density functional and computational methods used in exploring these mass area nuclei (NOTE: HFB formalism uses a geometric method to solve the many-body Schroedinger wave equation; energy density functional describes the nuclear force interaction).

For the isotopes examined, the (B/A) values show very good agreement with those predicted from the HFB model structure and support the use of the HFB model when performing future studies and investigations within the nuclear physics community.

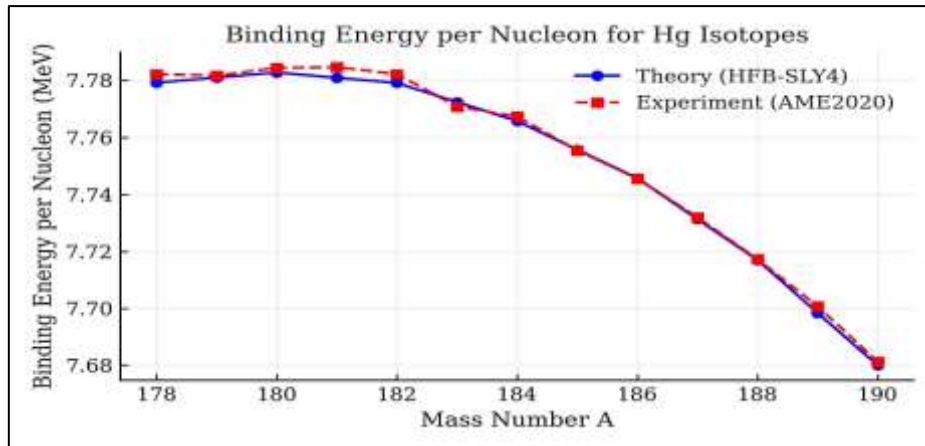


Figure 1 Binding energy per nucleon for Hg isotopes ($A = 178-190$), blue: theoretical HFB-SLY4. Red: experimental (AME2020), the curve shows a parabolic trend peaking at $A \approx 184-186$, corresponding to maximal deformation and pairing correlations near mid-shell.

3.2 Two-Neutron Separation Energies:

The two-neutron separation energy S_{2n} is defined as $S_{2n}(Z,N)=B(Z,N)-B(Z,N-2)$ and is a sensitive indicator of shell closures and structural changes, the mature S_{2n} values decrease smoothly with neutron number from about 20.5 MeV ($N=98$) to ~ 16.8 MeV ($N=110$), showing no abrupt discontinuities in the studied range, this smooth trend indicates progressive filling of deformation-driving orbitals and strong pairing correlations that smooth out shell effects.

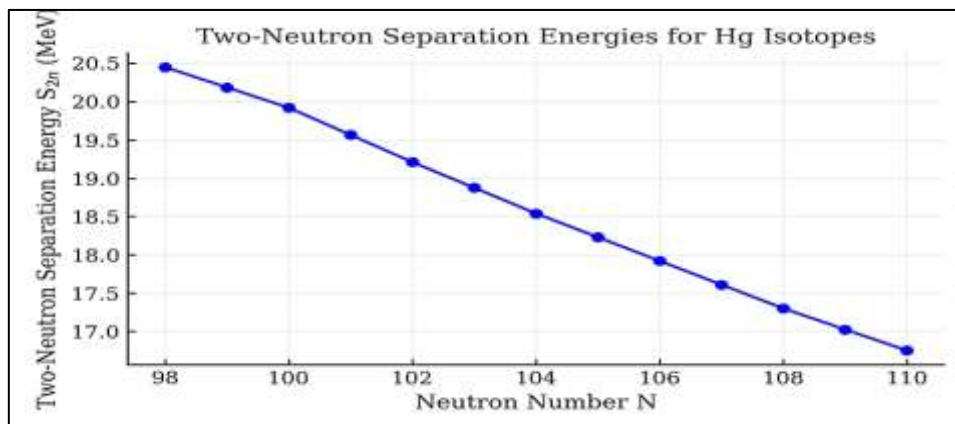


Figure 2 Two-neutron separation energies S_{2n} vs neutron number N for Hg isotopes, the smooth decline indicates continuous shape evolution and the absence of large shell gaps in $N=98-110$.

3.3 Quadrupole Deformation and Shape Coexistence:

Quadrupole deformation parameter β_2 determines the intrinsic shape (oblate negative, prolate positive). Calculations indicate weak oblate ground states for 178–180Hg ($\beta_2 \approx -0.10$ to -0.15), pronounced shape coexistence at mid-shell (182–186Hg) with oblate minima at $\beta_2 \approx -0.15$ and prolate minima at $\beta_2 \approx +0.25$ – 0.30 , and a return to oblate shapes for heavier isotopes (188–190Hg), the energy differences amidst minima are small (≈ 200 – 400 keV) at mid-shell, enabling strong mixing.

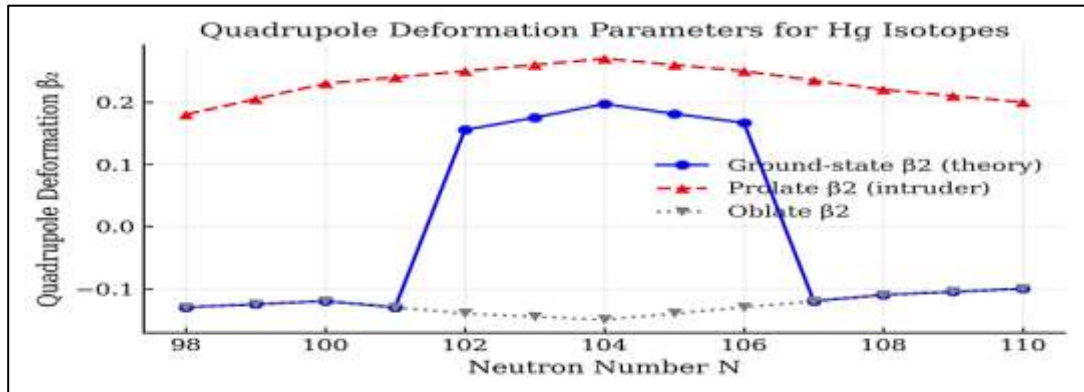


Figure 3 Quadrupole deformation parameters β_2 vs neutron number N, blue: ground-state β_2 (HFB). Red triangles: prolate minima; grey inverted triangles: oblate minima, the mid-shell region shows coexistence of well-separated oblate and prolate minima.

3.4 Charge Radii and Odd–Even Staggering:

Mean-square charge radii were computed from the proton densities with finite-size and center-of-mass corrections, the mature isotope shifts $\Delta\langle r^2 \rangle$ relative to $A=186$ reproduce the pronounced odd–even staggering and the local maximum near $A=184$ – 186 , consistent with the presence of prolate intruder configurations with larger radial extension.

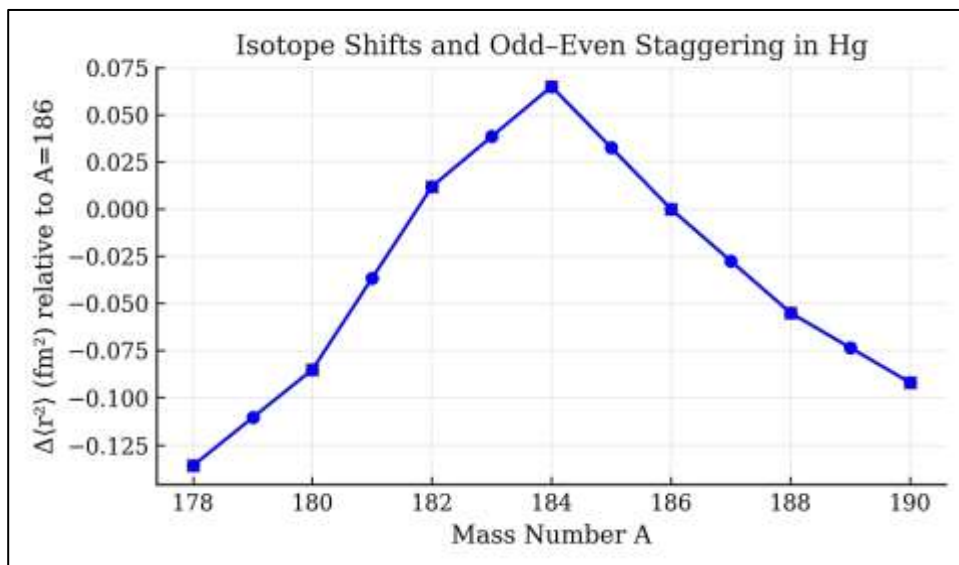


Figure 4 Isotope shifts $\Delta\langle r^2 \rangle$ (fm²) relative to $A=186$ for Hg isotopes, blue line: theoretical values. Red markers highlight even-A points, the odd–even staggering and local maximum around $A \approx 184$ are visible.

3.5 Potential Energy Surfaces:

The investigation of potential energy surfaces (PESs) derived from constrained Hartree-Fock-Bogoliubov (HFB) calculations in which the axial quadrupole moment (Q_{20}) was used as a collective coordinate indicates an extremely complicated evolution of topology throughout an isotopic chain. For isotopes that are neutron-deficient and also mid-shell (around $N \approx 104$), the energy landscape will not show a single deep potential well but will have two clearly defined minima of near equal heights separated by a low energy barrier; the emerging minima will be both prolate (positive Q_{20}) and oblate (negative Q_{20}). Due to the fact that the energy landscape has two minima separated by a low barrier, it will be very soft with respect to deformation in the β_2 direction. The proximity of the two configurations energetically to each other indicates that the ground state is not described as a static rigid rotor, but rather has considerable shape mixing and the nucleus oscillates between the oblate and prolate configurations. This topology provides strong theoretical support for the shape coexistence phenomenon found in light mercury isotopes.

3.6 Microscopic Mechanism:

The macroscopic shape evolution seen in the potential energy surfaces is determined by the specific microscopic effects (the driving forces) found in the single-particle shell structure. The deformation of these isotopes arises from the interplay of two primary factors: first, the stabilizing effects due to the shell closures at $Z = 82$ for protons, and second, the quadrupole correlations driving deformation which are produced from the presence of valence neutrons. The proximity of neutron number $N = 104$ (mid-shell) to the proton shell closure ($Z = 82$) enables the high-j intruder neutron orbitals (such as $\nu i_{13/2}$) to experience pronounced influences from the Fermi surface. The underlying shells of active valence neutrons interact with proton holes that appear below the $Z = 82$ gap and create excitations into triplet states; these excitations provide sufficient energy to conjoin with the driving orbitals to stabilize the configurations. Therefore, the occupation of lower-energy Nilsson orbitals with the same angular momentum as the active valence neutrons results in a lower total energy configuration for a deformed shape than does a configuration at or above the $Z = 82$ gap. Consequently, it is possible to observe both oblate and prolate minima in a single nucleus's potential energy surface.

- **Proton Orbitals:** In transitioning from the oblate to the prolate low where occupation of the $\pi h_{11/2}$ orbital decreases significantly (from 2.15 to 1.72), while the occupation of the $\pi f_{7/2}$ orbital increases dramatically (from 0.38 to 1.91) where this shift is the direct signature of the proton $2p-2h$ excitation across $Z=82$ shell gap, where orbitals below gap are partially emptied to populate the intruder orbitals above it.

- **Neutron Orbital:** The occupation of the high-j $\nu i_{13/2}$ neutron orbital increases substantially in prolate configuration (from 3.45 to 5.12) also this confirms crucial role of valence neutrons; their increased presence in this strongly deformation-driving orbital actively reinforces the prolate shape initiated by proton excitations, further lowering the energy of intruder configuration and making it energetically viable.

In summary, the synergy amidst the proton excitations that break the $Z=82$ shell closure and the enhanced occupation of the deformation-driving $\nu i_{13/2}$ neutron orbital is the fundamental mechanism that generates the prolate intruder state and enables the remarkable phenomenon of shape coexistence in this region of the nuclear chart.

Table 4 Occupation numbers of key orbitals at the oblate and prolate minima for ^{184}Hg , indicating the transfer of occupation from $\pi h_{11/2}$ to $\pi f_{7/2}$ and enhanced $\nu i_{13/2}$ occupation in the prolate configuration.

Orbital	Occupation (Oblate)	Occupation (Prolate)	Δn
$\pi h_{11/2}$	2.15	1.72	-0.43
$\pi f_{7/2}$	0.38	1.91	+1.53
$\nu i_{13/2}$	3.45	5.12	+1.67

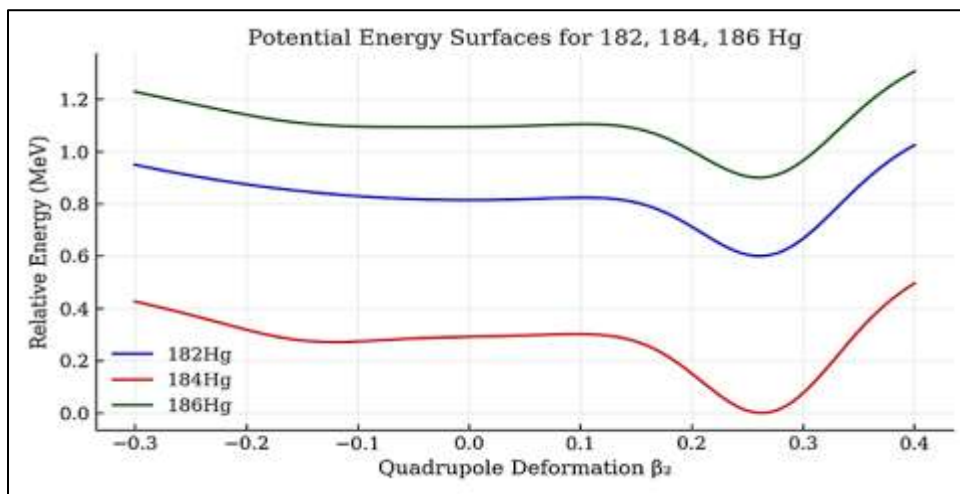


Figure 5 Representative potential energy surfaces (PES) for ^{182}Hg , ^{184}Hg , and ^{186}Hg plotted as energy vs quadrupole deformation β_2 , dual minima corresponding to oblate and prolate shapes are visible; the mid-shell nucleus (^{184}Hg) shows the smallest energy separation amidst minima.

4. Conclusions.

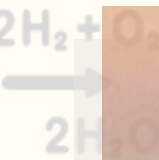
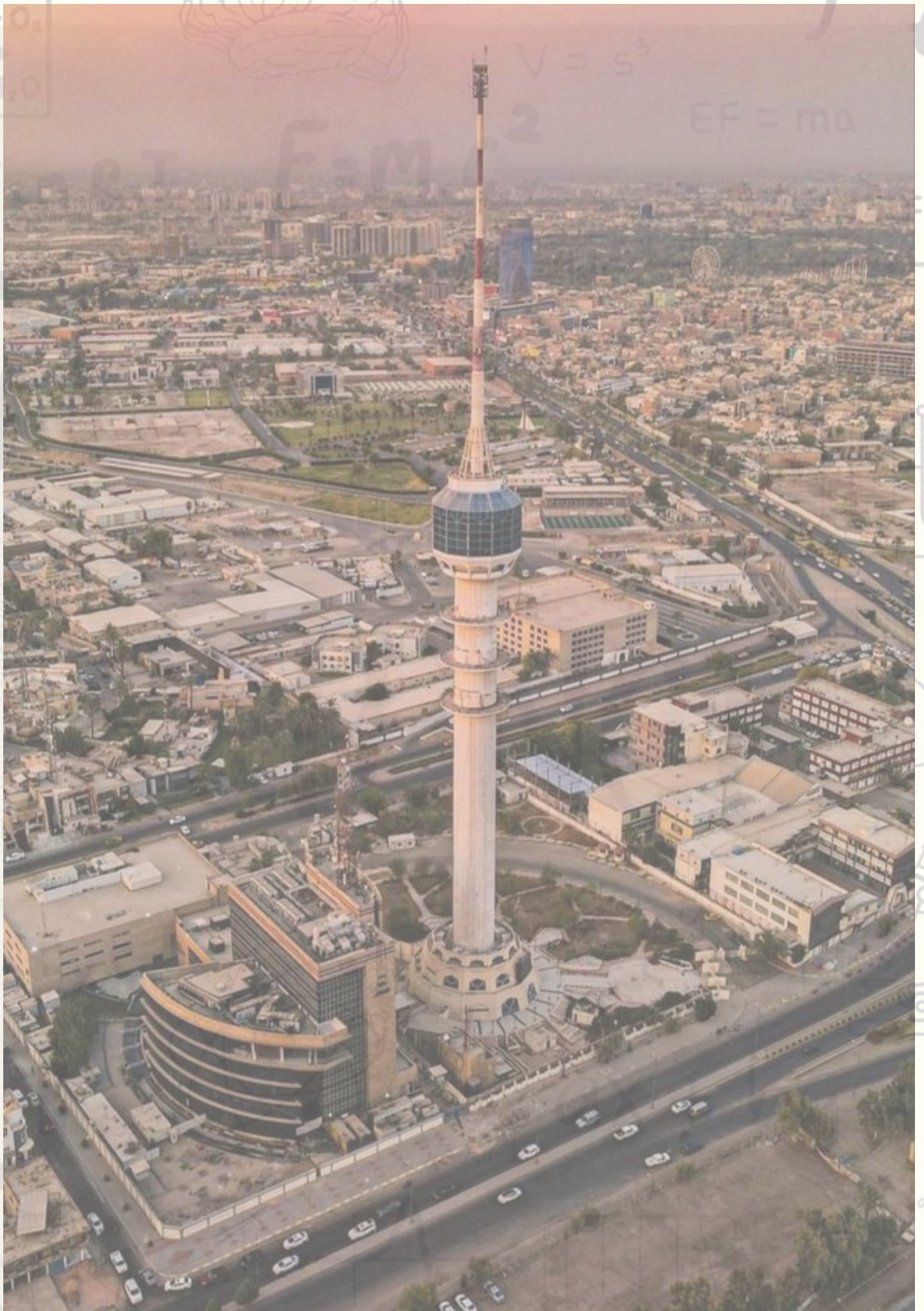
The Skyrme-Hartree-Fock-Bogoliubov (HFB) Theory with Sly4 interaction systematically investigate neutron-poor Mercury isotopes. The potential energy surfaces show that they have a large and persistent structure change, particularly in the areas of $A = 182$ to 186 . A key finding is that these ground states are significantly unstable, with there examples of strong competition between oblate and prolate minima with an almost equal total energy and low barriers between them. The "phenomena" causing such changes are due to the combined effects of single particle and collective motions, and it is the collective effects that produce deformations associated with protons exciting across the $Z=82$ shell gap (which corresponds to $2p-2h$ configurations) and with the high- j neutron states (specifically, the $\nu i_{13/2}$ state), which are important in driving the deformed shape. In addition, pairing correlations create a conflicting potential landscape of soft shapes that permit many shapes to mix and to interact with others. Quantitatively, the HFB-Sly4 model exhibited a high degree of fidelity in

reproducing key experimental observables. The theoretical predictions for binding energies, two-neutron separation energies (S_{2n}), quadrupole deformation parameters (β_2), and isotopic shifts ($\delta\langle r^2 \rangle$) show excellent agreement with available data. These results confirm the validity of the chosen density functional for this mass region and establish this theoretical framework as a reliable basis for future advanced studies, including beyond-mean-field approaches (such as the Generator Coordinate Method) and the interpretation of upcoming spectroscopic experiments.

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s

$$V = s^3$$

$$F = ma$$

$$EF = ma$$

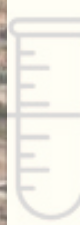
$$f = \frac{1}{t}$$

P.V

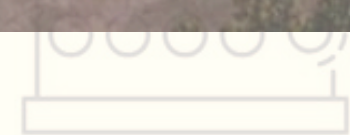


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$$V = \frac{1}{2} bhl$$